

A Systematic Review of Electrochemical Model-based Lithium-ion Battery State Estimation in Battery Management Systems

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Abstract

This study presents a systematic review of electrochemical model-based battery state estimation methods. A search was conducted in Web of Science, Scopus, and IEEE Explore databases, resulting in the selection of 102 relevant research articles. These articles were carefully analyzed to summarize research trends, model selection, and estimation methods. Research trends cover publication years, journals, keywords. In terms of model selection, the review examines scalability, programming languages & software, battery materials, the choice of models, model simplification techniques, and parameter identification processes. The estimation methods are assessed based on the specific battery states and the algorithms employed for state estimation. The results indicate that while research in this area has been growing, it primarily focuses on individual cells, with limited attention paid to battery modules and packs. There is a noticeable lack of standardization in model simplification and parameter identification processes. Most studies concentrate on a single state estimation, with state of charge being mostly investigated, followed by state of health and state of temperature. Future research should address state estimation for modules and packs, parameter identification, model simplification, joint multi-state estimation, and better utilization of internal battery information for improving battery performance and safety.

Keywords: Electrochemical models, Lithium-ion batteries, Battery state estimation, Battery management systems, State of charge, Systematic review

Abbreviations

AEM	Average-electrode model
BMS	Battery management systems
CKF	Cubature Kalman filter
DFN	Doyle-Fuller-Newman
ECMs	Equivalent circuit models
EKF	Extended Kalman filter
ESPM	Extended Single Particle Model
FDM	Finite difference method
FEM	Finite element method
FVM	Finite volume method
GA	Genetic algorithm
KF	Kalman filter
LCO	Lithium Cobalt Oxide
LFP	Lithium Iron Phosphate
LMO	Lithium Manganese Oxide
MCMB	Mesocarbon Microbeads

MHE	Moving horizon estimation
MPM	Many-Particle Model
NCA	Lithium Nickel Cobalt Aluminum Oxide
NMC	Lithium Nickel Manganese Cobalt Oxide
P2D	Pseudo-Two-Dimensional
PDE	Partial differential equations
PF	Particle filter
PI	Proportional-integral
PID	Proportional-integral-derivative
PRISMA	Preferred reporting items for systematic reviews and meta-analyses
PSO	Particle swarm optimization
RQ	Research questions
SEIKF	Singular evolutive interpolated Kalman filter
SMO-ERL	Sliding-mode observers with an exponential reaching law
SMO-URL	Sliding-mode observers with a uniform reaching law
SOB	State Of Balance
SOC	State Of Charge
SOE	State Of Energy
SOF	State Of Function
SOH	State Of Health
SOP	State Of Power
SOT	State Of Temperature
SOX	State Of X
SPM	Single Particle Model
SPMe	Single Particle Model with Electrolyte
UKF	Unscented Kalman filter

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41 1 Introduction

42 As the notion of carbon neutrality increasingly gains traction as a developmental objective among
43 nations worldwide, clean energy and energy storage technologies have witnessed remarkable progress
44 in recent years [1]. Lithium-ion batteries have emerged as a fundamental energy storage solution
45 across various applications, encompassing electric vehicles, portable electronics, and grid energy
46 storage. Owing to their high energy density, long cycle life, and comparatively minimal self-discharge
47 rates, they represent the preferred option for numerous applications [2]. To preserve optimal
48 performance, safety, and longevity, battery management systems (BMS) are utilized to supervise and
49 regulate battery operations. Consequently, the advancement of secure and dependable BMS has
50 become a focal point of research, and the establishment of such systems is of paramount importance
51 [3, 4].

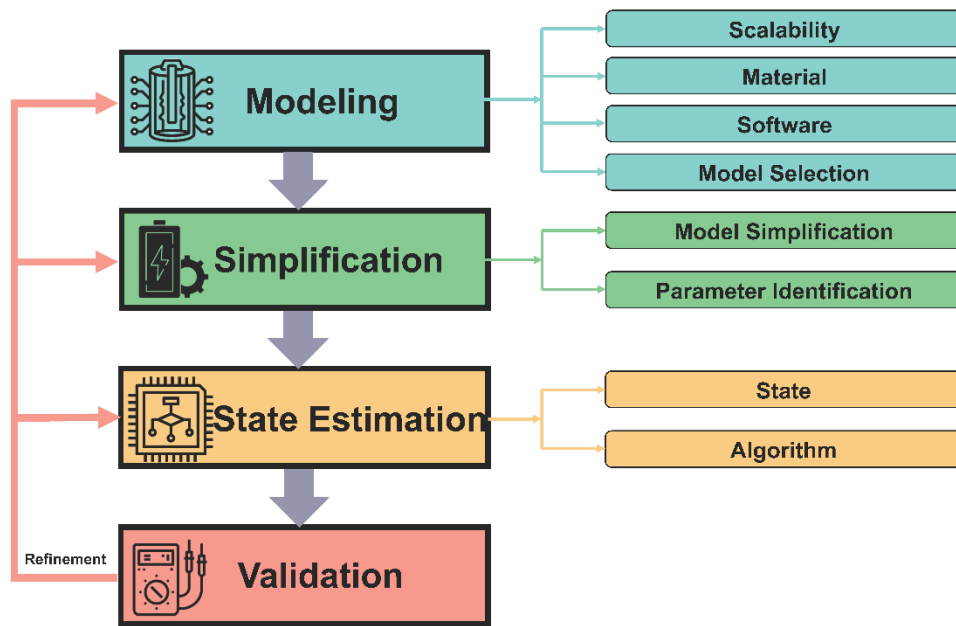
52 A main component of BMS involves estimating various battery states, including State Of Charge
53 (SOC), State Of Energy (SOE), State Of Health (SOH), State Of Power (SOP), State Of Temperature
54 (SOT), State Of Balance (SOB), State Of Function (SOF), among others. These battery states can be
55 collectively referred to as State of X (SOX) [4]. The SOC reflects the remaining capacity of the
56 battery and is the most central state for battery management and control [4]. SOE is similar to SOC,
57 except that it is defined in terms of residual energy [4]. The SOH reflects the health state of the
58 battery. Regarding SOH, there are two definitions, one based on the decay of battery capacity, and
59 one based on the increase of internal resistance [4]. The SOT of a battery is a crucial factor that
60 significantly influences its safety. Battery temperature can typically be categorized into three types:
61 battery surface temperature, battery core temperature, and battery bulk temperature. Battery core
62 temperature is usually the maximum temperature inside the battery, and battery bulk temperature
63 refers to the average temperature inside the battery [4]. SOP is an important state that affects the
64 safety and performance of a battery, and SOP reflects how much maximum power a battery can
65 output at a given moment [5]. The SOB is an emerging concept in battery research, focusing on the
66 assessment and regulation of uniformity within a battery system [6]. SOF is an indicator used to
67 determine whether a battery system can meet the requirements of a specific application [4].
68 Accurately determining these states is crucial for ensuring battery safety and optimizing both battery
69 life and performance. The aforementioned battery states are often not directly measurable and
70 necessitate estimation based on battery voltage, current, temperature, and additional information
71 gathered through sensors and historical operating data. Lithium-ion batteries function as chemical
72 energy storage systems. Their internal reaction mechanisms are inherently complex and nonlinear.
73 This complexity makes the accurate description and estimation of various battery states particularly
74 challenging [7].

75 Battery state estimation techniques primarily encompass direct measurement methods [8],
76 machine learning methods [9], and model-based methods [10-13]. The direct measurement method
77 like Coulomb counting has the advantage of simple logic and does not require a large amount of
78 computation, but often results in large errors due to initial values and measurement noise. Machine
79 learning approaches adeptly address nonlinear problems and are well-suited for battery state
80 estimation applications. Nevertheless, machine learning techniques function as black boxes,
81 precluding an understanding of the internal logic, and this makes it very difficult to perform fault
82 diagnosis. At the same time, bottlenecks in computation capacity and data storage limit the
83 application of machine learning methods. Model-based approaches typically exhibit commendable
84 estimation accuracy and robustness when applied in the BMS. However, the accuracy of model-based
85 techniques heavily relies on the accuracy of the battery model [11]. Designing a model-based battery
86 state estimation method based on a highly accurate battery model could yield superior estimation
87 accuracy and enhance BMS performance.

88 The most common model-based methods used to estimate battery states in BMS rely on
89 equivalent circuit models (ECMs). These models are characterized by their minimal computational
90 complexity and therefore can be accommodated by BMS processors [14]. However, ECMs are
91 phenomenological models, hence they cannot provide an accurate estimation of certain battery states,
92 such as SOH, owing to an insufficient understanding of the influence on future behaviors, aging, and
93 safety levels. Consequently, this can lead to the design of overly conservative estimation algorithms,
94 which cannot fully exploit the existing potential of the battery. Additionally, the electrochemical

95 model offers a greater advantage over the ECMs by providing more detailed information about the
 96 battery internals. This is more beneficial for enhancing BMS performance and ensuring battery safety.
 97 Physics-based models rooted in fundamental electrochemical principles offer notable advantages. One
 98 prominent example is the Pseudo-Two-Dimensional (P2D) model, also referred to as the Doyle-
 99 Fuller-Newman (DFN) model or Newman model [15]. It describes the battery dynamics with a set of
 100 coupled partial differential equations (PDE). Employing the electrochemical battery model facilitates
 101 a more comprehensive understanding of the battery's internal information. Nonetheless, employing
 102 electrochemical models for the development of battery management systems encounters a multitude
 103 of obstacles, including the necessity for simplification of electrochemical models owing to the
 104 computational capacity constraints of current BMS processor. Furthermore, designing battery state
 105 estimation algorithms based on intricate electrochemical models presents a formidable challenge.
 106 Consequently, conducting a review of the research advancements in this domain is of paramount
 107 importance.

108 The flowchart of battery state estimation based on electrochemical models is presented in Fig. 1.
 109 The process commences with the modeling, which includes modeling for a single cell or a battery
 110 pack, the specific battery materials, the choice of software or programming language, and the
 111 selection of the precise electrochemical model. The second step entails model simplification,
 112 specifically involving the simplification of the electrochemical model and the identification of model
 113 parameters. The third step comprises the state estimation using the electrochemical model, including
 114 the choice of states to estimate and the selection of the specific estimation method. Finally, the
 115 validation.



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 117 **Fig. 1. Battery state estimation based on electrochemical models.**
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119 Currently, there are a number of published literature reviews in this field. Park et al. [16]
 120 conducted an extensive analysis of contemporary approaches to battery state estimation methods
 121 through a literature review. Nevertheless, a significant proportion of the articles examined primarily
 122 focus on the ECMs methodology. Comprehensive reviews of battery modeling and state estimation
 123 approaches are presented by Wang et al. [4] and Hu et al. [17], although their assessments only
 124 cursorily address the electrochemical model and fail to offer an in-depth exploration of battery state
 125 estimation based in the electrochemical model. Although Planella et al. [18] provide a detailed outline
 126 of the electrochemical battery models, their investigation does not include battery state estimation
 127 techniques. As a result, a systematic review centering on state estimation based on electrochemical
 128 models remains absent.

129 The systematic literature review is a novel approach to review literature [19]. Traditional
 130 literature reviews tend to be limited by the author's understanding and bias in retrieving papers.

131 Unlike traditional literature review methods, it provides a systematic method for literature retrieval
132 and writing process. This approach significantly reduces research biases. Additionally, it ensures the
133 presentation of results is both comprehensive and transparent. In this paper, we followed the process
134 in Fig. 1 to present a systematic literature review that encapsulates battery state estimation methods
135 based on electrochemical models, equipping researchers with valuable references and insights for
136 future studies. The specific research questions (RQ) to be addressed are as follows:

137 RQ1: What were the research trends?

138 RQ2: What electrochemical battery models were used for battery state estimation?

139 RQ3: Which battery state was selected for estimation and what algorithms were used?

140 **2 Method**

141 This section will refer to the PRISMA method [19] for systematic literature reviews. It
142 specifically includes: literature search strategy, literature selection criteria, and data extraction process.
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144 **2.1 Literature search strategy**

145 This study searched a total of three high-quality databases, namely: Web of Science, Scopus and
146 IEEE Explore. The following Boolean logic were searched: (“battery” OR “batteries”) AND
147 (“electrochemical” OR “physic*”) AND (“state estimation” OR “state of charge” OR SOC OR “state
148 of energy” OR SOE OR “state of health” OR SOH OR “state of power” OR SOP OR “state of
149 temperature” OR SOT OR “state of balance” OR SOB OR “state of function” OR SOF). The search
150 keywords are divided into three main levels: battery, electrochemical model, and battery state. The
151 search keywords are searched in the title, keywords and abstract of the article. To reduce the work of
152 screening papers in next steps, during the search process, filters are applied to select only journal
153 articles, English-language articles, and to exclude literature reviews for a more focused and efficient
154 search. The retrieval time range is up to December 31, 2023.
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156 **2.2 Selection criteria**

157 The papers were screened using the PRISMA method. The specific process is shown in Fig. 2. In
158 the first step, the papers were searched in the three databases using the method introduced in Section
159 2.1, and a total of 6597 articles (Web of Science:2715, Scopus:2819, IEEE Explore:1063) were
160 retrieved. The first step was performed to de-duplicate the papers, leaving 3455 articles remaining.
161 The second part is to screen the articles according to their titles and abstracts, and a total of 2375
162 articles are screened out in this step. The third step is to check whether the full text of the article is
163 available. This step is all available in full text, so no articles are excluded. The fourth step is to read
164 the full text of the article and make a detailed screening according to the screening criteria, which are
165 shown in Table 1. It is worth noting that, although the search filters were set to exclude review articles
166 and non-English articles, some such articles were still retrieved. This is due to errors in the database
167 system's labeling. After the fourth step of screening, all articles that meet the requirements can be
168 ultimately obtained. Finally, 102 articles [20-121] were selected to meet the requirements (see
169 Appendix 1).

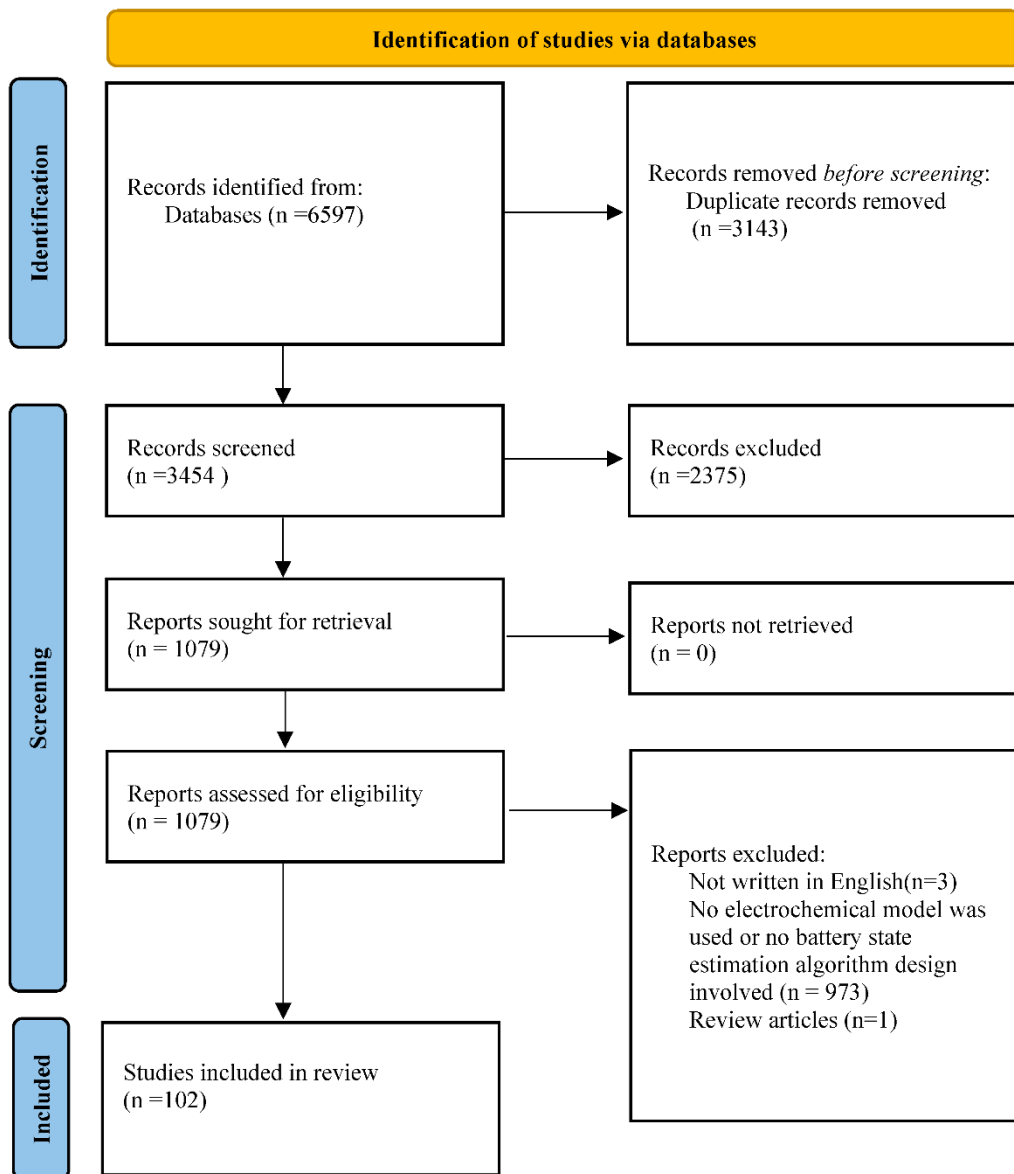


Fig. 2. Article Screening Results.

Table 1 Exclusion and Inclusion Criteria.

Inclusion Criteria	Exclusion Criteria
Lithium-ion battery	Solid-state or other types of batteries
Using P2D and its derivative battery models	Not using P2D and its derivative battery models
Battery state estimation algorithm design using electrochemical model	No electrochemical model was used, or no battery state estimation algorithm design involved
Lithium-ion battery	Other types of batteries
Journal Papers	Non-journal papers
Empirical study	Literature Review
Written in English	Written in other languages

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176 **2.3 Data extraction process**

177 The data selection process was guided by three research questions outlined in the Introduction
 178 section. Data extraction will be carried out from Table 2, i.e.

- 179 1. Extraction centered on research trends: including trends in number of papers published
 180 annually, distribution of published papers across journals, distribution of keywords in papers.
 181 2. Extraction focused on electrochemical models: battery formation methods, utilized
 182 programming languages and software, types of battery materials employed, electrochemical
 183 model utilized, model simplification method employed, and method used for identifying
 184 battery model parameters.
 185 3. Extraction pertained to state estimation methods: which the battery states are estimated, and
 186 the techniques used to estimate them.
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Table 2 Data extraction.

Category	Item	Description
RQ1: Research trends	Years	Number of papers published annually
	Journals	Distribution of published papers across journals
	Keywords	Distribution of keywords in papers
RQ2: Electrochemical models	Scalability	Battery formation methods for single cells, modules, or packs
	Programming Language & Software	Utilized programming languages and software
	Materials	Types of battery materials employed
	Model Selection	Electrochemical model utilized
	Model Simplification	Model simplification method employed
	Parameter Identification	Method used for identifying battery model parameters
RQ3: State estimation	State	Battery states estimated
	Algorithm	Estimation methods used

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191 **3 Results**

192 Following the data extraction method introduced in Section 2.3, the specific results are presented
 193 in detail in this section.
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195 **3.1 RQ1: What were the research trends?**

196 This subsection answers the first research question and delves into the research trends in
 197 electrochemical model-based battery state estimation. Specifically, it examines the publication year,
 198 the academic journals, and the specific keywords utilized within these screened articles.
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200 **3.1.1 Year**

201 Fig. 3 illustrates the annual publication trend of research articles focusing on battery state
 202 estimation using electrochemical models. Commencing in 2006 with the paper of Santhanagopalan et
 203 al. [74], a steady growth in the number of such studies is evident. Notably, there has been a marked
 204 upsurge in pertinent publications since 2020, signifying an escalating interest among researchers
 205 within this domain. Especially in 2023, the number of publications called a greater increase in the
 206 previous years, indicating that the battery state estimation based on electrochemical model has
 207 attracted more and more researchers' attention.

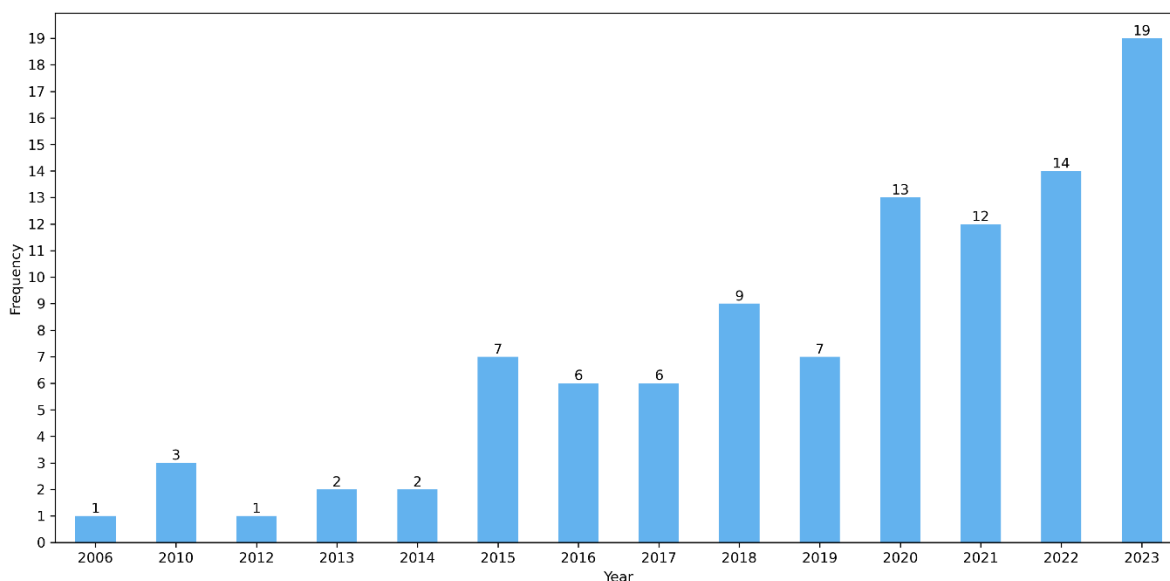


Fig. 3. Annual paper publication trends.

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3.1.2 Journals

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Fig. 4 demonstrates the distribution trend of published articles in this field. Journals with less than 2 publications are classified as Others. Research in this domain is predominantly featured in high-impact factor journals specializing in electrochemistry, control, and energy. This can be attributed to the interdisciplinary nature of the subject, necessitating an amalgamation of expertise in electrochemistry, control, and energy applications. The leading three journals contributing to this area of research include Journal of Power Sources (17), IEEE Transactions on Control Systems Technology (14), and Journal of Energy Storage(10).

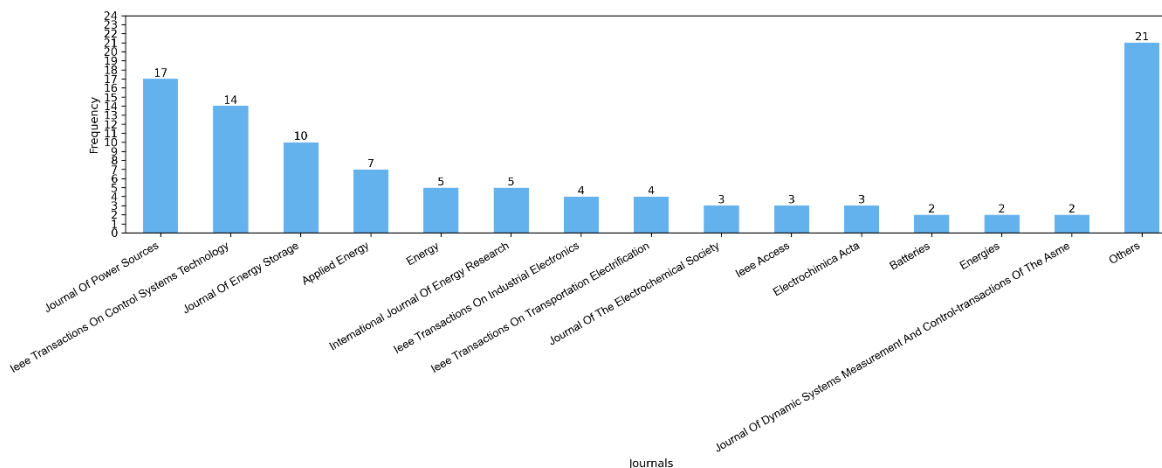


Fig. 4. Trends in journal publication.

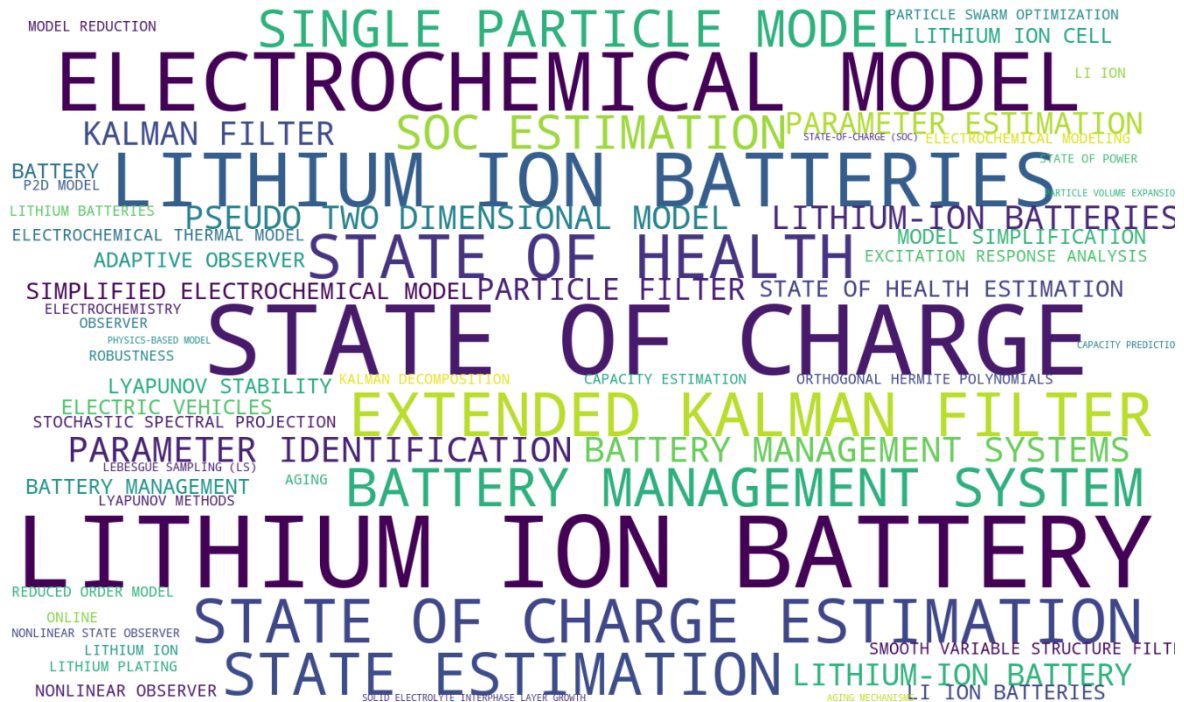
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3.1.3 Keywords

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In this subsection, a keyword analysis of the 102 selected articles is conducted, noting that 8 (ID1, ID12, ID27, ID30, ID41, ID48, ID49, ID72) of these articles did not provide keywords. Therefore, a total of 94 articles that provided keywords are analyzed. A word cloud figure is employed to illustrate the frequency of the keywords. Fig.5 exhibits the results, with larger font sizes signifying higher frequencies of the corresponding keywords. This visualization offers a rapid understanding of the primary themes and concepts underpinning the research.

233 "Lithium-ion battery" and "Lithium-ion batteries" are the most frequently employed keywords,
 234 while "Electrochemical model" is also frequently mentioned. "State of charge", "State of charge
 235 estimation" and "SOC estimation" are the most commonly cited battery state keywords, followed by
 236 "State of health". The "Pseudo-two-dimensional model" and the "Single particle model" are the two
 237 most prevalent electrochemical models. "Extended Kalman filter" and "Kalman filter" are the most
 238 widely used methods for battery state estimation. The multiple mentions of "Battery management
 239 system" and "Battery management systems" indicate researchers' interest in the application of battery
 240 state estimation methods within battery management systems. "Parameter identification" also appears
 241 with high frequency, highlighting its importance in model-based battery state estimation. The
 242 accuracy of the battery model parameters greatly influences the final estimation results, underscoring
 243 the significance of identifying the parameters of the battery model.
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 246 **Fig. 5. Word cloud for keywords.**
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249 **3.2 RQ2: What electrochemical battery models were used for battery state estimation?**

250 This subsection answers the second research question. Specifically, scalability, programming
 251 language and software, battery material, model selection, model simplification and parameter
 252 identification.
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254 **3.2.1 Scalability**

255 Scalability, in the context of the use of battery, typically encompasses three levels [122]: cell,
 256 module, and pack. A cell is the fundamental unit, which is connected in series or parallel
 257 configurations to create a module. Subsequently, multiple modules are interconnected in series or
 258 parallel arrangements to form a comprehensive battery pack. This hierarchical structure exemplifies
 259 the scalability concept in battery formation. The statistical results on the battery state estimation for
 260 each battery formation level are shown in Table 3.

261 Of the 102 articles analyzed, all focused on cell-level state estimation algorithms. A mere 4
 262 studies extended the cell-level battery state estimation to encompass both module and pack levels.
 263 Article ID 37 investigates SOC estimation for a battery pack comprising 6 single cells connected in
 264 series. Similarly, article ID 73 explores SOC estimation for a battery pack containing 4 single cells
 265 connected in series. And article ID 53 examines SOC and SOT for a battery pack with 12 single cells
 266 connected in series. ID87 uses electrochemical model to study the battery SOC and SOT estimation
 267 for the case where the cells are connected in series or in parallel. Most research focuses on cell-level

268 because modeling and state estimation for single cell is relatively straightforward. The models are
 269 easier to construct, and simulations and experiments are simpler to conduct. In contrast, battery packs
 270 and modules consist of multiple cells connected in series or parallel, leading to more complex
 271 electrical and thermal coupling effects, which make state estimation significantly more challenging.
 272 These findings suggest that most of the current research in this domain is centered on cell-level state
 273 estimation, with minimal investigation into module-level and pack-level state estimation.
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Table 3 Battery formation results.

	Number	Article ID
Cell	102	All Articles
Module or pack	4	ID37, ID53, ID73, ID87

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 277 **3.2.2 Programming language and software**

278 In this study, the usage of programming languages and software in the analyzed papers is
 279 quantified. The articles mentioned the programming language or software utilized, with the results
 280 presented in Table 4.

281 It is evident that MATLAB (57) is the most frequently employed software, primarily because it is
 282 highly suited for control algorithm development and facilitates the design of battery state estimation
 283 algorithms. Moreover, MATLAB offers a comprehensive library of functions that streamline battery
 284 modeling and parameter identification.

285 COMSOL Multiphysics (19) is the second most used software, with its battery simulation results
 286 typically serving as benchmarks or points of comparison to verify the accuracy of proposed models or
 287 battery state estimation methods. LabVIEW (2) is generally utilized in hardware-in-the-loop testing,
 288 while Fortran (2), C++ (1), and C (1) are employed for battery simulation to reduce the calculation
 289 time.

290 Given the complexity of electrochemical modeling, some open-source battery models can be
 291 leveraged as a foundation for battery state estimation research. Examples of such models and their
 292 associated programming languages include Slide (C++) [123], DUALFOIL (Fortran) [124], fastDFN
 293 (MATLAB) [125], LIONSIMBA (MATLAB) [126], and PyBaMM (Python) [127].
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Table 4 Programming language and software

Programming language and software	Number	Article ID
MATLAB	57	ID1, ID3-ID7, ID9-ID11, ID13, ID14, ID16-19, ID21-23, ID25, ID27, ID29, ID30, ID31, ID33, ID40, ID41, ID43, ID46, ID51, ID53, ID57, ID59, ID60, ID62, ID63, ID65-71, ID73, ID74, ID83-90, ID92, ID97-100
COMSOL	19	ID7, ID11, ID22, ID23, ID25, ID28, ID31, ID39, ID43, ID44, ID53, ID56, ID59, ID66, ID68, ID71, ID85, ID90, ID99
LabVIEW	3	ID13, ID57, ID73
Fortran	2	ID12, ID71
C++	1	ID30
C	1	ID30

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 297 **3.2.3 Materials**

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 299 The active materials in the battery define the voltage boundaries and the shape of the cell voltage
 300 curve. Since mostly carbon based materials are used as negative electrode the battery types are named
 301 after the cathode material [128].

302 Of the 103 articles analyzed, 87 articles mention the battery materials used in their modeling.
 303 Currently, the most researched battery types in this field are Lithium Cobalt Oxide (LCO) (25),
 304 Lithium Iron Phosphate (LFP) (25), and Lithium Nickel Manganese Cobalt Oxide (NMC) (24).
 305 Lithium Manganese Oxide (LMO) (9) and Lithium Nickel Cobalt Aluminum Oxide (NCA) (6) are

306 also prevalent lithium battery cathode materials. LMO-NMC signifies a combination of LMO and
 307 NMC materials. All the papers report graphite or Mesocarbon Microbeads (MCMB) as the negative
 308 electrode materials. Table 5 provides a list of the corresponding papers for each positive electrode
 309 material for reference. Since many of the parameters in electrochemical modeling are related to their
 310 materials, most of the papers in Table 5 list or cite the model parameters used and can be used as a
 311 reference for the reader.

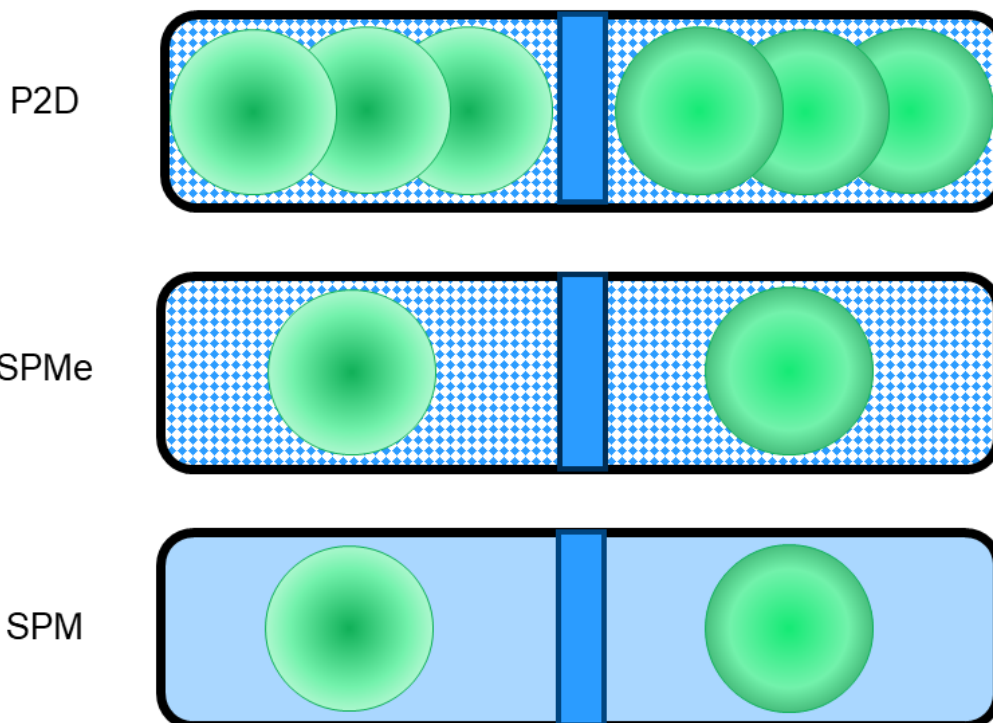
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Table 5 Positive electrode materials

Positive electrode materials	Number	Article ID
LCO	25	ID5, ID14, ID21, ID29, ID35, ID36, ID40, ID43, ID46, ID52, ID55, ID60, ID61, ID63, ID67, ID69-73, ID78, ID87, ID94, ID98, ID102
LFP	25	ID4, ID7-9, ID18, ID28, ID31, ID32, ID37, ID39, ID42, ID48, ID49, ID59, ID65, ID74, ID75, ID79, ID81, ID86, ID88, ID89, ID96, ID100, ID101
NMC	24	ID1, ID11, ID13, ID17, ID19, ID23, ID25, ID30, ID33, ID34, ID39, ID53, ID54, ID57, ID58, ID77, ID80, ID82, ID84, ID85, ID89, ID93, ID97, ID99
LMO	9	ID2, ID6, ID15, ID22, ID38, ID44, ID53, ID56, ID66
NCA	6	ID39, ID53, ID64, ID83, ID89, ID91
LMO-NMC	3	ID26, ID62, ID76

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3.2.4 Model selection



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Fig. 6. Electrochemical Model.

321 This subsection provides an overview of the battery models employed in the screened papers.
 322 Based on our screened literature, electrochemical battery models can be broadly categorized into three
 323 main types as shown in Fig. 6. The first category is the P2D model, which serves as the foundational
 324 framework for all other electrochemical models. The P2D model provides a comprehensive
 325 description. However, when this model is simplified by reducing the multiple particles to a single
 326 representative particle, it is referred to as the Single Particle Model with electrolyte (SPMe). Further
 327 simplification occurs in the Single Particle Model (SPM), where the concentration diffusion process
 328 in the electrolyte is disregarded, and an averaged concentration value is used instead. These
 329 hierarchical models—P2D, SPMe, and SPM—reflect varying levels of complexity and approximation,
 330 offering different trade-offs between computational efficiency and accuracy depending on the specific
 331 application requirements. The results are presented in Table 6. Notably, all battery models are
 332 simplifications based on the P2D model. If that are not specifically named in the papers, are then
 333 categorized under the P2D category.

334 Out of the total, 47 articles utilized SPM, which offer significant simplification, enabling the
 335 design of battery state estimation algorithms with reduced computational effort. The SPM [129] is a
 336 simplified model, positing that a lithium-ion battery consists of a single representative particle for
 337 each electrode (cathode and anode), neglecting the spatial distribution of potential and concentration
 338 within the electrodes. The average-electrode model (AEM) (ID3, ID10, ID32, ID41, ID42), a type of
 339 the SPM model that replaces both lithium-ion concentration and current density with average values.
 340 Instead of a single representative particle, the AEM considers the entire electrode's average behavior.
 341 Consequently, the AEM captures the spatial distribution of lithium-ion concentration and potential
 342 within the electrodes, offering a better understanding of battery performance and efficiency. As a
 343 result of this simplification, the SPM model is less accurate during high-current charging and
 344 discharging. To address this, SPMe has been proposed.

345 Twenty-five papers have utilized the SPMe. It is worth noting that different studies may use
 346 varying terminologies, such as Extended Single Particle Model (ESPM) (D14, ID25, ID33, ID54,
 347 ID58, ID59, ID62, ID79, ID80, ID84, ID85). However, these models are fundamentally based on the
 348 SPM with additional physical equations to describe the electrolyte dynamics [130]. Meanwhile, ID74
 349 proposes a battery model considering the electrical double layer effect and performs battery state
 350 estimation using this model. ID81 employed Many-Particle Model (MPM). MPM differs from SPMe
 351 in that there is not just one particle in the electrode, but multiple particles.

352 Thirty articles employed the P2D model as a basis for simplification when designing battery state
 353 estimation algorithms. Due to the varying simplifications used in most papers, and the lack of specific
 354 model names, they are collectively classified as P2D models. Some authors named their P2D model as
 355 1-dimensional models. ID24 employs a model proposed by Subramanian et al. [131], simplifying it
 356 through a constant electrolyte concentration and an approximate solution to the diffusion equation.
 357 ID70 applies the model proposed by Forman et al. [132].

358 The aforementioned results outline the battery models used in the analyzed papers. However,
 359 these models often require further simplification before being applied to battery state estimation. The
 360 following subsection examines the model simplification methods employed in these articles.
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Table 6 Battery model types

Model	Number	Article ID
SPM	47	ID1-5, ID9, ID10, ID12, ID15, ID17, ID18, ID20, ID28, ID29, ID32, ID35-38, ID40-42, ID48-ID52, ID55, ID57, ID65-68, ID70, ID72, ID73, ID76-ID78, ID82, ID87, ID88, ID94-96, ID98, ID100
SPMe	25	ID8, ID13, D14, ID21, ID22, ID25, ID26, ID30, ID33, ID44, ID54, ID56, ID58, ID59, ID62, ID74, ID79, ID80, ID81, ID84, ID85, ID91, ID93, ID101, ID102
P2D	30	ID6, ID7, ID11, ID16, ID19, ID23, ID24, ID27, ID31, ID34, ID39, ID43, ID45-47, ID53, ID60, ID61, ID63, ID64, ID69, ID71, ID75, ID83, ID86, ID89, ID90, ID92, ID97, ID99

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3.2.5 Model simplification

Electrochemical models encompass numerous PDE, which present considerable complexity in solving and are unsuitable for real-time computations. Consequently, battery model simplification is typically necessary for battery state estimation. In particular, it is important to note that the accuracy of a simplified model will necessarily be reduced. Model simplification is a matter of finding a balance between model accuracy and model complexity. As illustrated in the selected articles, various methods exist for battery model simplification, namely: simplification of physical processes, finite method and function approximation, spectral method, transfer function, projection method and Artificial intelligence (AI) method. Since many of the screened articles directly use other researchers proposed models or do not mention the specific process of model simplification, this section only summarizes articles that explicitly propose model simplification methods. The results are displayed in Table 7.

Table 7 Simplification approach

Method	Number	Article ID
Finite method	27	FDM: ID3, ID9, ID10, ID15, ID16, ID17, ID20, ID27, ID30, ID35, ID38, ID41, ID46, ID49, ID51, ID52, ID54, ID60, ID65, ID77, ID81, ID95 FVM: ID71, ID97 FEM: ID50, ID90, ID99
Function approximation	37	Parabolic approximation : ID4, ID5, ID14, ID18, ID21, ID22, ID24, ID32-34, ID36, ID37, ID40, ID42, ID47, ID53, ID55, ID64, ID66-69, ID71, ID72-75, ID78, ID79, ID80, ID84, ID88, ID89, ID93, ID94, ID96, ID101
Spectral method	6	Chebyshev polynomial: ID6, ID43, ID59, ID71, ID86, ID99
Transfer function	22	Padé approximation: ID1, ID2, ID12, ID13, ID19, ID25, ID26, ID29, ID44, ID45, ID52, ID56, ID57, ID70, ID76, ID85, ID91, ID92, ID100 Discrete realization algorithm: ID31, ID52 Fractional-order functions: ID82
Projection method	4	ID7, ID23, ID52, ID83
AI method	1	ID102

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(1) Simplification of physical processes

Simplification of physical processes means directly ignoring some reaction processes considered as unimportant. For instance, the SPM reduces the electrode to a particle, while the AEM employs the average value to represent the lithium-ion concentration and current density. Most articles adopt methods that simplify physical processes, most frequently by converting a distribution to a constant or average value. This approach effectively streamlines the model and minimizes computational effort while maintaining model accuracy within certain boundaries like low C-rate. In ID11 and ID97, the lithium diffusion in the solid phase is simplified by equivalent circuit model. This simplifies the lithium diffusion physical processes within the cell, thus making the model simpler.

The approach of simplifying physical processes is very convenient to implement and allows for the simplification of specific parts as needed, striking a good balance between model complexity and accuracy. Unlike the methods discussed later, which focus on how to solve PDE and enhance the speed and accuracy of their solutions. Simplification of physical processes can be combined with the methods introduced later to further simplify the model.

393 **(2) Finite method**

394 Finite method refers to the technique of converting a continuous spatial domain into a finite set of
395 discrete points or elements, which is crucial for solving PDE that characterize a wide range of
396 physical and engineering phenomena. The finite difference method (FDM) and the finite volume
397 method (FVM) are frequently employed. The finite element method (FEM) is another well-known
398 approach. The advantage of finite method is that it can be easily applied, and it generally used as
399 benchmark methods to solve PDE. But the accuracy of the finite method is related to how many nodes
400 it uses for computation, the more nodes you have the more accurate results you can get, but it also
401 means longer computation time and larger memory requirements.

402 For finite method, there are also methods to reduce the amount of computation while ensuring
403 high accuracy. Simplification using the theory of largescale systems, an FDM containing 50 nodes
404 can be simplified to contain only 5 nodes(ID3, ID10). In ID 16, a second-order Runge-Kutta method
405 to solve an FDM with 10 node. But as the number of FDM nodes increases, the solution of Runge-
406 Kutta method will be unstable. ID97 uses the Implicit-Explicit Method to discretize the FVM, which
407 is typically faster and more accurate to solve compared to Implicit Euler Method.

408 The advantage of the finite method is that it is very easy to apply, and the number of nodes can be
409 adjusted according to the actual needs to balance the accuracy and computational speed. It is possible
410 to combine different discretization methods to speed up the computation. The finite method is often
411 used as a baseline method for solving PDE.

412 **(3) Function approximation**

413 The function approximation method is a widely adopted approach for simplifying complex
414 models by replacing computationally or expressively complex parts with simpler functions. The most
415 common function approximation method used in electrochemical model is the parabolic
416 approximation method [131]. Based on the results in Table 7, it can be seen that the number of papers
417 employing the parabolic approximation method is the greatest. This is because it's very easy to apply.
418 And the calculation is very fast, which is crucial for application in BMS. At the same time this
419 method also retains the physical meaning and can be used to directly calculate the average lithium-ion
420 concentration and the surface lithium-ion concentration, which are two state quantities that are
421 important for battery state estimation. However, it has the disadvantage of lower accuracy than other
422 methods, especially at currents greater than 1C [133]. At the same time, the parabolic approximation
423 method cannot add nodes to further improve the accuracy like finite method or spectral method.
424 Nevertheless, if the goal is to develop battery state estimation algorithms that can be applied in a BMS,
425 it is also a good choice due to its computational speed.

426 **(4) Spectral method**

427 Spectral methods are an efficient way to solve PDE and are also used to solve PDE in
428 electrochemical models. In spectral methods, the function to be solved is expressed as a set of basic
429 functions, with Chebyshev polynomials being the most common basis function. All retrieved papers
430 using spectral methods employ Chebyshev polynomials as basis function. Spectral methods have been
431 proven to generally offer higher accuracy and faster convergence rates compared to finite methods.
432 Furthermore, accuracy can be further improved by increasing the number of nodes. However, a
433 disadvantage is that they are complex to implement and require a strong background in mathematics,
434 which is why only a few papers have applied this method.

435 **(5) Transfer function**

436 The transfer function method uses the Laplace transform, which converts a function of a real
437 variable t (usually time) into a function of a complex variable s (complex frequency). The transform
438 has the useful property of converting differential equations to algebraic equations and simplifying the
439 model. When the PDE is converted to the frequency domain, it can be solved using different methods.
440 The most commonly used solution method is the Padé approximation. There are also a very few
441 screened papers that use discrete realization algorithm and fractional-order functions for PDE solving.
442 The transfer function approach has the advantage of high accuracy and the choice of different orders
443 of approximation to balance accuracy and computational speed. At the same time, choosing a lower
444 order for the approximation may cause the model to perform poorly in the high frequency part.

445 **(6) Projection method**

446 The projection method is a mathematical technique used to solve PDE. This method simplifies the
447 problem solving process by finding approximate solutions by projecting an infinite-dimensional

448 problem into a finite-dimensional subspace. Similar to spectral methods, projection methods can also
 449 be complex to implement in programming. Consequently, fewer papers employing projection methods
 450 were identified in the screened literature.

451 **(7) AI method**

452 AI method can also be used to solve PDEs. In ID102, the authors introduced a novel approach
 453 termed the Physics-Informed Multiple-Input Operator Network (PI-MIONet) for solving PDEs within
 454 electrochemical models. Unlike traditional numerical methods typically employed for PDE resolution,
 455 this AI-driven model leverages the principles of physics to inform the network's learning process. The
 456 findings demonstrated that PI-MIONet is not only capable of accurately solving the PDEs inherent in
 457 electrochemical models but also offers a potentially more efficient and scalable alternative to
 458 conventional numerical techniques. This advancement underscores the growing potential of AI
 459 models in complex scientific computations, particularly in fields where traditional methods may be
 460 computationally intensive or challenging to implement.

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 462 **3.2.6 Parameter identification**

463 The accuracy of the battery model is directly affected by the correct selection of the battery model
 464 parameters. The battery model parameters also have a significant impact on the accuracy of the
 465 model-based state estimation results. The identification of the parameters of the electrochemical
 466 model is relatively complicated, and in the research papers on electrochemical model-based state
 467 estimation method, the battery parameters are generally adopted from the existing papers. However,
 468 even for the same model and type of battery, the battery parameters of each battery may be slightly
 469 different due to manufacturing and other reasons. So, to improve the accuracy of the battery state
 470 estimation algorithm, it is necessary to re-identify some parameters of the battery model. Among the
 471 102 articles screened, 27 articles mentioned battery parameter identification, and Table 8 summarizes
 472 them.

473 Within the scope of these 27 articles, particle swarm optimization (PSO) is the most used methods
 474 for parameter identification. The majority of these papers focus on re-identifying a limited number of
 475 parameters in the model, while still relying on data from other publications for the remaining
 476 parameters. Notably, only ID59 and ID84 re-identify over 20 parameters. Additionally, ID93 is also a
 477 good choice to perform parameter sensitivity analysis first and then select high sensitivity are
 478 parameters for estimation.

479 Operating conditions employed for parameter identification generally fall into two categories:
 480 constant current charging and discharging, and dynamic operating conditions. Most articles
 481 exclusively utilize either constant current conditions or dynamic conditions. This is because it is
 482 simpler to use only one operating condition for battery parameter identification. However, ID59, ID60,
 483 and ID84 stand out by incorporating both types of conditions for battery parameter identification.
 484 Employing a more comprehensive range of conditions yields more accurate battery parameters.
 485 Conversely, relying on a single condition increases the likelihood of the algorithm identifying only a
 486 local optimum rather than a global optimum solution.

487 Temperature exerts a significant influence on battery model parameters. A mere 4 papers (ID1,
 488 ID4 ID77, and ID84) consider experimental battery data at multiple temperatures for parameter
 489 identification. The remaining papers consider only a single temperature. Broadly speaking,
 490 incorporating a greater variety of experimental data conditions and temperature conditions contributes
 491 to obtaining more precise battery model parameters.

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Table 8 Summary of battery parameters identification method

Article ID	Approach	Identified Parameters	Experimental Data	Ambient Temperature
ID1	Adaptive particle swarm optimization	6	0.1C discharge	0°C, 25 °C, 45°C
ID4	least square optimization	16	-	0 °C, 15 °C, 25°C, 35°C, 55°C
ID5	PSO	12	0.2C discharge	20°C

ID16	GA	7	DST	25 °C
ID17	PSO	4	-	-
ID18	GA	-	-	-
ID20	Levenberg– Marquardt algorithm	-	CCCV (1C) charge	25 °C
ID36	PSO	-	-	-
ID38	Levenberg– Marquardt algorithm	7	-	-
ID49	least square optimization	7	-	-
ID51	least square optimization	6	-	-
ID58	GA	16	HPPC	23 °C
ID59	Ant Lion Optimizer	21	2C discharge US06 DST UDDS	25 °C
ID60	Gradient free function minimization algorithm	9	A set of four charges (1 A, 2A and 3A at constant current and 4A with pulse current demand) and four discharges (2 A, 3.5A and 5A at constant current and 6A pulse current demand)	-
ID62	Nelder-Mead	-	1C charging/discharging	-
ID74	GA	6	0.1C charging/discharging	25 °C
ID77	GA	7	0.05C, 0.33C, 0.5C, and 1C charging/discharging	5°C,15°C,25°C,35°C,45°C,55°C
ID79	Linear decreasing weight particle swarm optimization	6	From 0.1 to 2C charging/discharging	25 °C
ID84	PSO	26	US06, DST and BJDST starting from different SOC levels	0°C,25°C,45°C
ID86	PSO	11	UDDS	25 °C
ID88	PSO	3	1C, 2C, 3C charging/discharging	-
ID91	PSO	-	UDDS	-
ID93	PSO	8	-	-
ID94	PSO	11	-	Room temperature
ID96	PSO	6	1C, 2C charging/discharging	25 °C

ID99	PSO	6	0.5C, 0.7C, 0.9C, 1C, 1.5C charging/discharging	-
ID101	Multi-verse optimizer (MVO) algorithm	15	-	-

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3.3 RQ3: Which battery state was selected for estimation and what algorithms were used?

This subsection shows which battery states researchers have focused on and what methods they have used to state estimation.

3.3.1 State

Fig. 7 provides an overview of the battery states estimated in the included articles. A total of 83 articles addresses the estimation of SOC, demonstrating that the majority of research concentrates on this particular aspect. The SOH ranks second, with 26 articles examining concerns related to battery aging and lifespan. Additionally, 11 articles investigate the temperature estimation (SOT) of the battery, typically focusing on both the internal core temperature and the surface temperature. A smaller subset of studies, comprising 4 articles, employs electrochemical models to estimate the SOP of the battery. Furthermore, 2 articles utilize electrochemical models to analyze SOB. In summary, the SOC and SOH are the two battery states that currently garner the most attention from researchers. SOE and SOF are not mentioned by any article. The specific results can be found in Appendix 1.

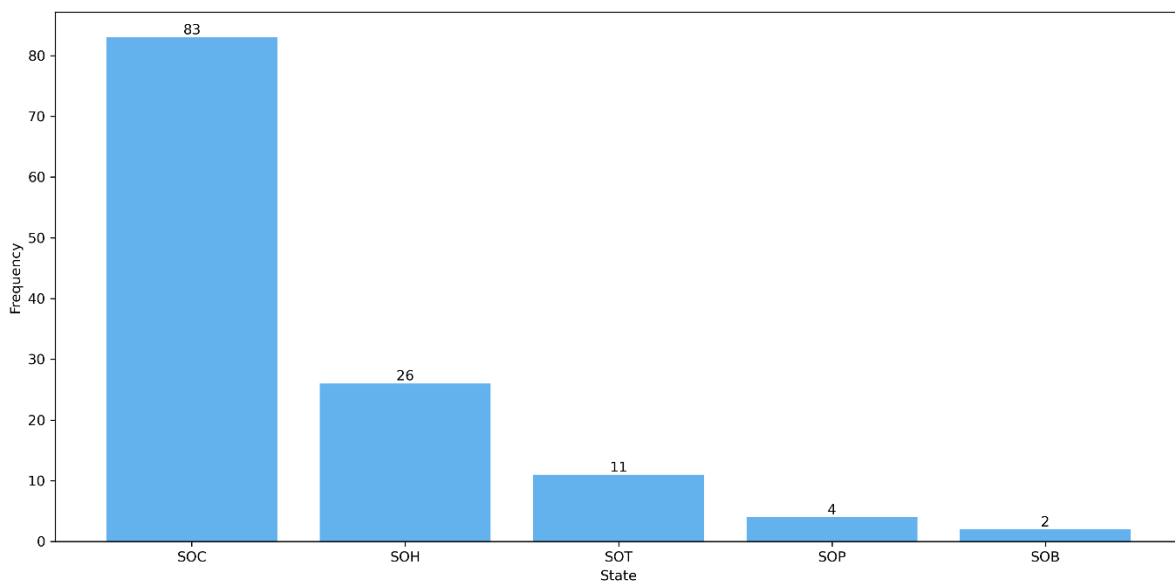
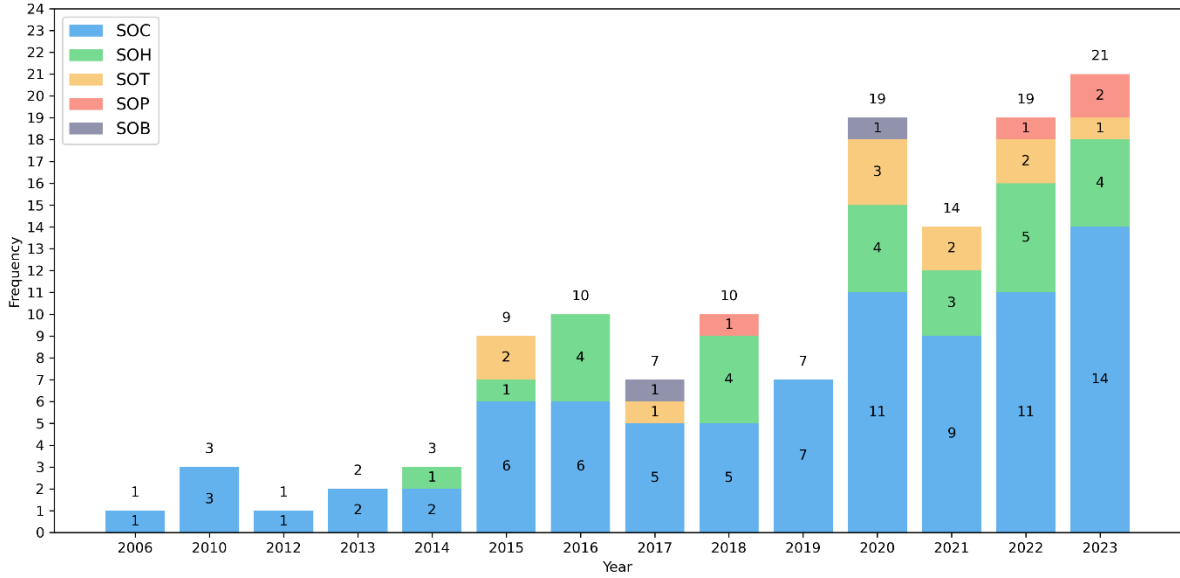


Fig. 7. Battery state statistics results.

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Fig. 8 illustrates the yearly distribution of battery states, revealing that the SOC consistently remains the predominant focus across all years. In recent times, there has been a noticeable increase in the number of researchers concentrating on the SOT, SOP, and SOH. As these three states are intimately connected to battery safety, this shift in research interest underscores a growing emphasis on monitoring and ensuring the safe usage of the battery. It is worth noting that the sum of the battery states discussed above exceeds the total number of articles. The reason for this discrepancy is that many of the articles involve estimates for more than one battery state. From Appendix 1, a total of 80 articles focused on estimating a single battery state, indicating that the majority of the research is centered around one specific state, typically either the SOC or SOH. Furthermore, 20 articles estimated two battery states concurrently, with the most prevalent combinations being SOC&SOH or SOC&SOT. A mere two articles (ID10 and ID11) estimated three battery states simultaneously. ID10

525 estimates SOC, SOT, and SOB while ID11 estimates SOC, SOT, and SOH. Notably, no papers were
 526 found to estimate more than three battery states at once.



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 529 **Fig. 8. Battery state results distributed by year.**
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532 3.3.2 Algorithms

533 In this subsection, we provide a detailed account of the specific methods employed for battery
 534 state estimation in the reviewed literature for each battery state.

535 (1) SOC

536 In electrochemical models, the SOC is typically represented by calculating the normalized
 537 lithium-ion concentration in electrodes. This method provides a quantitative measure of the SOC by
 538 comparing the current concentration of lithium-ions to the maximum possible concentration, thus
 539 indicating the battery's charge level. The SOC is a critical parameter for assessing the performance
 540 and efficiency of the battery, and it is determined using the following equation:

$$541 \text{SOC} = \frac{C_{Li} - C_{Li,min}}{C_{Li,max} - C_{Li,min}} \quad (1)$$

542 where C_{Li} represents the average lithium-ion concentration in the electrode, $C_{Li,min}$ is the minimum
 543 lithium-ion concentration, and $C_{Li,max}$ is the maximum lithium-ion concentration. This equation
 544 normalizes the concentration between 0% and 100%, providing a standardized measure of the
 545 battery's charge state. Typically, the SOC of the anode is used to represent the overall SOC of the
 546 battery. In some cases, the average SOC of both the anode and cathode is employed to provide a more
 547 comprehensive indication of the battery's SOC. Therefore, SOC estimation algorithms based on
 548 electrochemical models primarily focus on estimating the average lithium-ion concentration (C_{Li}) in
 549 both the anode and cathode. This estimated average lithium-ion concentrations are then used to
 calculate the overall SOC of the battery by Eq. (1).

550 There are three primary categories of SOC estimation methods based on electrochemical models:
 551 model calculation method, filter, observer and AI method. The specific SOC estimation techniques
 552 implemented in each paper are summarized in Table 9.

553 **1) Model calculation method:** The model calculation method does not involve designing filters
 554 or observers for SOC estimation. Instead, it directly utilizes electrochemical models for computation,
 555 typically calculating the lithium-ion concentration and subsequently deriving the battery's SOC. The
 556 advantage of this method is that it is straightforward, but it cannot handle the influence of
 557 measurement noise and unknown initial conditions, which makes it difficult to guarantee high
 558 accuracy in practical applications in BMS. To overcoming the effects of initial value errors, ID73 and
 559 ID89 use the OCV method in the initial phase of the algorithm to compute the correct initial value,

560 and then use this correct initial value to initialize the battery model. This solves the problem of the
561 initial value, but it also cannot deal with the interference caused by measurement noise.

562 **2) Filters:** The second primary category for SOC estimation encompasses various types of filters.
563 The primary purpose of a filter is to estimate the state of a system in the presence of noise. It not only
564 relies on the current measurements but also integrates previous state estimates and the system's
565 dynamic model. A filter effectively reduces the impact of measurement noise, thereby providing a
566 more accurate state estimation.

567 The Kalman filter (KF) is the most commonly used filter. The original KF requires that the
568 dynamic and measurement processes be linear. However, electrochemical models are often nonlinear,
569 making it necessary to adapt the KF to handle nonlinear systems. As a result, variations of the KF that
570 can accommodate nonlinearity are commonly employed in the design of filters based on
571 electrochemical models. The most widely used variations include the Extended Kalman Filter (EKF),
572 the Unscented Kalman Filter (UKF), and the Cubature Kalman Filter (CKF). Among these, the EKF
573 is the most frequently employed option, often serving as the benchmark algorithm for comparison
574 purposes. Several studies (ID10, ID26, ID27, ID32, ID41, ID42, ID43, ID45, ID47, ID52, ID55, ID60,
575 ID67, ID68, ID71, ID72, ID74, ID79, ID81 and ID98) have utilized the EKF. Numerous filter
576 variations based on the EKF have also been implemented, such as the dual EKF in ID19 and ID100,
577 adaptive EKF in ID33, ID37 and ID67, Lebesgue-sampling-based EKF in ID36 and ID94, EKF
578 combined with smoothing variable structure filter in ID3. The UKF has been applied in ID18, ID23,
579 ID31, ID47, ID64, ID72, ID75, and ID102, while adaptive UKF filter is used in ID25, dual UKF in
580 ID46, adaptive square-root sigma-point KF in ID13. Sigma-point KF and UKF are the same filter with
581 different name. There are also other KF-based variations that have been used for SOC estimation,
582 such as: singular evolutive interpolated Kalman filter (SEIKF) in ID11, cubature Kalman filter (CKF)
583 in ID84, adaptive CKF in ID80, and square-root CKF in ID59. The EKF linearizes the system model
584 by performing a first-order Taylor expansion of the nonlinear system. Specifically, EKF uses the
585 Jacobian matrix to approximate the nonlinear functions of the system with linear ones. The advantage
586 of EKF is its relatively low computational complexity. However, its drawback is that it can lead to
587 significant estimation errors in highly nonlinear systems. The UKF utilizes the Unscented
588 Transformation to directly propagate the mean and covariance, capturing the key characteristics of the
589 distribution by selecting a set of "sigma points" in the state space, without requiring linearization.
590 Therefore, UKF typically provides better estimation accuracy than EKF, although at the cost of
591 increased computational complexity. The CKF generally performs better when dealing with high-
592 dimensional systems, making it particularly well-suited for situations where there are many states to
593 estimate.

594 Another prominent filter type is the particle filter (PF), which has been employed in ID26, ID34,
595 ID64, ID86, and ID93. The advantage of the PF is its ability to directly handle nonlinear systems;
596 however, its computational complexity is higher than that of the KF. Some studies have also explored
597 other filters for SOC estimation, such as moving-window filtering in ID7, State-dependent-Riccati-
598 equation filter in ID9, and smooth variable structure filter in ID65.

599 **3) Observers:** The observers are another common method for SOC estimation. The observer is
600 designed to estimate the internal state of a system by measuring its output. Compared to filters,
601 observers generally have lower computational complexity, as they primarily rely on the system model
602 for feedback adjustments and do not need to handle large amounts of noise and uncertainty.

603 Nonlinear observers are utilized in ID22, ID29, ID50, ID51 and ID87, ID91, ID95, while ID5
604 employs a nonlinear observer with terminal voltage feedback injection. ID40 uses a feedback observer.
605 Nonlinear adaptive observers are employed in ID17, ID49, and ID70. ID24 features an output error
606 injection observer, whereas ID12 and ID78 incorporate backstepping PDE state observers. ID20 and
607 ID56 utilize proportional-integral (PI) observers, and ID44 opts for a proportional-integral-derivative
608 (PID) observer. Luenberger observers are employed in ID69, with ID66 integrating a Luenberger
609 observer combined with a Recursive Least Squares method. ID21 makes use of moving horizon
610 estimation (MHE), and both ID32 and ID42 employ sliding-mode observers with a uniform reaching
611 law (SMO-URL) as well as sliding-mode observers with an exponential reaching law (SMO-ERL).
612 Lastly, ID28 incorporates observers with the backstepping method and H-infinity observers.

613 **4) AI method:** AI methods have also been employed for SOC estimation based on
614 electrochemical models. In ID18, a neural network was added to the SPM model to correct its errors,

615 and a UKF was designed based on this hybrid model to perform SOC estimation. A combination of
 616 GA and electrochemical model is used in ID92 for battery SOC estimation. In ID99, the internal state
 617 information from the electrochemical model is used as input to train a neural network model for
 618 estimating SOC. In ID102, a neural network was used to solve the PDEs in the electrochemical model,
 619 and a UKF was designed based on this hybrid model to estimate the battery's SOC. The studies
 620 mentioned above indicate that AI methods show potential in state estimation based on electrochemical
 621 models.

622 **Table 9 SOC estimation method in corresponding paper results**

Method	Article ID
Model Calculation	ID1, ID4, ID14, ID15, ID35, ID53, ID58, ID63, ID64, ID73, ID83, ID89,
Filters	ID3, ID7, ID9-11, ID13, ID18, ID19, ID23, ID25-27, ID30-33, ID36, ID37, ID41-43, ID45-47, ID52, ID55, ID59, ID60, ID64-68, ID71, ID72, ID74, ID75, ID79-81, ID84, ID86, ID93, ID94, ID98, ID100, ID102
Observers	ID5, ID12, ID17, ID20-22, ID24, ID28, ID29, ID32, ID40, ID42, ID44, ID49-51, ID56, ID66, ID69, ID70, ID78, ID87, ID91, ID95
Machine learning	ID18, ID92, ID99, ID102

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626 **(2) SOH**

627 The electrochemical model can describe the physical process of battery aging well, which will be
 628 beneficial for SOH estimation. The SOH estimation methods used in the screened articles can also be
 629 classified into 4 categories, and the specific results are presented in Table 10.

630 **1) Model calculation method:** ID1 employs Gaussian process regression in conjunction with an
 631 SPM to calculate the remaining battery capacity. ID2, ID8, ID88, and ID96 develop an SPM-based
 632 battery aging model, which is then used to calculate the battery capacity loss. Empirical data was used
 633 to establish the relationship between battery capacity and the number of cycles in ID16. This was
 634 achieved by applying a third-order polynomial equation. ID35 implements a back-propagation neural
 635 network to estimate battery capacity and subsequently integrates the updated capacity into an SPM to
 636 compute the capacity-based SOH and the internal resistance-based SOH. The final SOH is determined
 637 as the minimum of these two values. Finally, ID39, ID61 and ID90 develop a battery aging model
 638 based on the P2D model. Most papers model battery aging and then directly apply the model's
 639 calculations to the aging process. This is equivalent to an open-loop estimation of SOH.

640 **2) Filters:** ID11 estimates the SOH utilizing the SEIKF. ID19 applies DEKF for SOH estimation.
 641 ID26 employs the Kalman smoother, PF, and EKF to estimate SOH. ID34, ID57, and ID62 utilize the
 642 PF for SOH estimation purposes. ID47 implements both the EKF UKF for estimating SOH. Lastly,
 643 ID52 and ID76 employ the EKF to estimate the SOH of the battery. ID101 uses PF to estimate SOH.

644 **3) Observers:** ID12 employs a backstepping PDE state estimator for SOC estimation, and the
 645 SOH-related battery model parameters are estimated using the least squares method in conjunction
 646 with the SOC state estimation results. ID20 utilizes a PI observer to estimate the battery capacity and
 647 resistance. ID21 implements a MHE technique for SOH estimation. ID40 applies an adaptive
 648 observer-based SOH estimator by estimating the maximum lithium-ion concentration of anode. ID54
 649 uses an adaptive interconnected observer that adjusts itself based on the relationship between the
 650 system's components for better capacity estimation. Lastly, ID82 incorporates an iterative model-
 651 based observer to estimate the internal resistance and capacity fade, which iteratively refines the
 652 estimation process to improve the accuracy of the SOH estimation.

653 **4) Machine learning:** ID35 uses neural networks to estimate capacity and then update the battery
 654 model.

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Table 10 SOH estimation method in corresponding paper results

Method	Article ID
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Model Calculation	ID1, ID2, ID8, ID16, ID35, ID39, ID61, ID88, ID90, ID96
Filters	ID11, ID19, ID26, ID34, ID47, ID52, ID57, ID62, ID76, ID101
Observers	ID12, ID20, ID21, ID40, ID54, ID82
Machine learning	ID35

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(3) SOT

As illustrated in Table 11, a total of 11 articles focusses on estimating the SOT. ID10 develops a thermal model, which calculates both core and surface temperatures. ID53 utilizes a lumped temperature model to estimate the temperature. ID11 employs SEIKF to determine the bulk temperature. ID18 applies UKF to estimate the battery core temperature, while ID43 uses EKF to estimate the bulk temperature. ID46 leverages dual UKF to estimate the bulk, surface, and core temperatures. ID59 implements square-root CKF for temperature estimation, and ID28 uses an H-infinity temperature observer to estimate the surface temperature. ID49 applies an adaptive observer for bulk temperature estimation, and ID78 utilizes a backstepping state observer to estimate the bulk temperature. ID87 uses a nonlinear observer to estimate the surface temperature.

The battery surface temperature can typically be measured using cost-effective temperature sensors, which renders the need for estimating the battery surface temperature less crucial. According to the findings in ID46, in certain situations, the battery core temperature can be as much as 45 °C higher than the surface temperature. Relying solely on bulk temperature estimation makes it challenging to prevent thermal runaway in the battery. Therefore, it is essential to estimate the core temperature of the battery to ensure better safety management.

Table 11 SOT estimation method in corresponding paper results

Method	Article ID
Model Calculation	ID10, ID53
Filter	ID11, ID18, ID43, ID46, ID59
Observer	ID28, ID49, ID78, ID87

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(4) SOP and SOB

A total of 4 articles (ID38, ID77, ID85, ID97) deal with the estimation of SOP, and there are fewer studies on SOP estimation based on electrochemical models. ID38 has calculated the SOP based on the lithium-ion concentration in the SPM, rather than using the commonly used terminal voltage and current. ID77 uses the grey wolf optimizer method for SOP estimation of batteries based on SPM. ID85 uses Gaussian process regression method to simplify the model and improve computational speed. Moreover, it proposes the concept of safe operation area (SOA), using the Electrode surface concentration, electrolyte concentration, slide-reaction overpotential and terminal voltage as indicators to set a safe operating range, and uses this as a constraint for the estimation of SOP. ID97 uses nonlinear model predictive control method for SOP estimation.

ID10 employs an active equalization control strategy based on the SPM to compute the battery's SOB. Meanwhile, ID48 adopts a multivariable SOC/capacity balancing strategy, utilizing the ESPM as a foundation for calculating the battery's SOB.

4 Discussion

The discussion section delves into the current issues and future research directions faced by electrochemical model-based, as well as the limitations of this study.

4.1 Issues and Future research

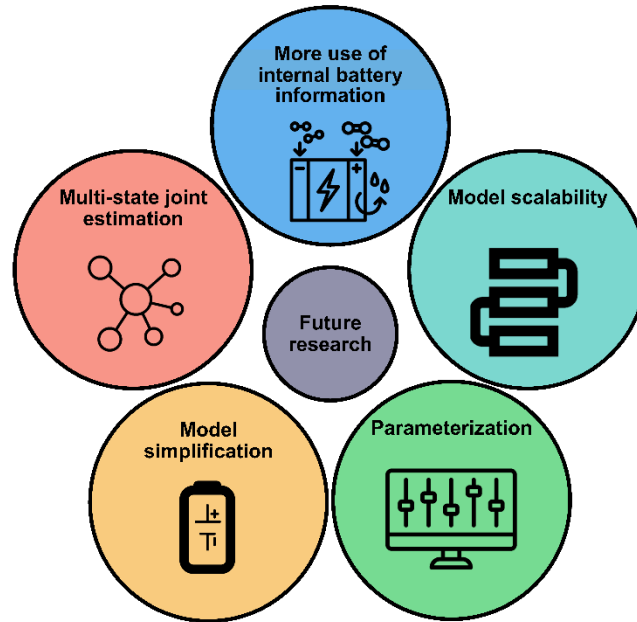


Fig.9. Future Research.

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Based on the results of Section 3, the current challenges and future research directions for battery state estimation based on electrochemical models are presented in Fig. 9, which specifically include: model scalability, parameterization, model simplification, multi-state joint estimation, and more use of internal battery information.

4.1.1 Model Scalability

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As can be seen from the results of Section 3.2.1, most existing studies focus primarily on individual cell. However, in practical applications, individual cells are often connected in series and parallel to form battery modules or packs. Due to manufacturing processes, achieving perfect consistency among new single cells is difficult. As cells are used, inconsistencies between them tend to increase for various reasons, such as discrepancies in manufacturing, uneven temperature distribution within battery modules and packs, and other factors.

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This disparity indicates that treating the entire pack as a singular large battery cell may compromise battery safety assurance. Similarly, the state of a battery pack cannot be unequivocally deduced by merely amalgamating the states of its constituent cells. The rationale behind this lies in the understanding that the overall performance of a battery module or pack may not be contingent upon the summative performance of all individual cells. However, adopting an electrochemical model-based estimation for each cell unit poses its own set of challenges, particularly regarding data storage and computational capacity of processors. So, the state estimation of batteries after aggregation is also very worthy of research. We can choose to estimate only a few battery cells in the most critical positions, thereby reducing the computational load on the BMS processor. This approach also provides guidance for the SOA. Consequently, it is suggested that forthcoming research should accentuate the state estimation for battery modules and packs based on electrochemical models. ID87 studies the state estimation of battery cells in series and parallel configurations based on an SPM model, which can provide a reference for subsequent research.

4.1.2 Parameterization

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Parameter identification for electrochemical models is very challenging due to the fact that the measurement of many parameters requires high precision equipment, and some parameters can't be measured directly. Furthermore, electrochemical models tend to be over-parameterized, meaning that two distinct sets of parameters may yield similar outputs. This complicates parameter estimation and raises the research threshold for studies based on electrochemical models. A notable issue is that many researchers do not provide complete battery parameters and test data in their publications.

733 In the future, further public disclosure of electrochemical model parameters and corresponding
734 test data would be advantageous, as it could lower the research barrier for electrochemical models.
735 This would not only benefit researchers working on battery state estimation and control based on
736 electrochemical models but also facilitate cross-sectional comparisons of various methods.
737 Additionally, there is currently a lack of comparative research on parameter identification methods,
738 such as comparing the accuracy, computational speed, and stability of different methods. For instance,
739 when performing parameter identification on batteries intended for secondary use, where each battery
740 exhibits varying degrees of aging, it becomes necessary to identify parameters for each individual cell.
741 If this process must be applied to 1000 cells and the chosen method is time-consuming, the total
742 duration can become excessively lengthy. Therefore, it is crucial to identify a parameter identification
743 method that not only ensures accuracy and stability but also operates with high computational
744 efficiency.

745 As observed in Section 3.2.6, there is a lack of standardized processes for parameter identification
746 methods in battery models, with different researchers employing diverse methods, operating
747 conditions, and temperatures. Parameter identification for electrochemical models often necessitates a
748 comprehensive understanding of the model's operating mechanism, as well as sensitivity analysis,
749 parameter grouping, and optimal experimental designs [134]. It would be worthwhile to propose a
750 simple and practical battery parameters identification process. This will effectively lower the
751 threshold for the use of electrochemical models.

752 **4.1.3 Model simplification**

753 Based on the findings presented in Section 3.2.5 and Section 3.2.6, it is evident that numerous
754 battery models and associated simplification methods have been developed with the goal of battery
755 state estimation. Statistically, the adoption of SPM-based models is by far the most prevalent.
756 However, when it comes to further simplification of these models, researchers have employed diverse
757 approaches, and there is a notable lack of cross-sectional comparisons. In particular, the lack of
758 discussions regarding which simplification methods are suitable for designing battery state estimation
759 algorithms. It must also be clear what limitations the simplifications imply like steady state
760 (dis)charging, low C-rate or narrow temperature band. If the application is targeting a BMS, then a
761 comparison of model computation speeds is also critical. The results of ID53 indicate that an
762 electrochemical model can be implemented in the BMS and offers higher accuracy than the ECM
763 while only adding an acceptable amount of operating time. Given these circumstances, the
764 investigation of model simplification for battery state estimation is an essential area for future
765 research. A more comprehensive understanding of the most effective simplification methods will
766 enable the development of improved battery state estimation algorithms, facilitating the application of
767 electrochemical models in BMS. Additionally, according to the study in ID102, using AI methods for
768 the simplification of electrochemical models, or developing hybrid models that combine AI with
769 electrochemical models, are also promising research directions.

770 **4.1.4 Multi-state joint estimation**

771 As indicated by the findings in Section 3.3.1, current research primarily focuses on the estimation
772 of individual battery states, with researchers designing specific estimation algorithms for each single
773 state. However, in practical applications, different states are often interconnected, such as the
774 estimation of SOP, which requires information on SOC. Compared with single-state estimation, multi-
775 state joint estimation methods are more advantageous in optimizing the efficiency and accuracy of the
776 overall algorithm.

777 Using an electrochemical model can get detailed internal battery information. This facilitates the
778 direct computation of some states based on their physical definitions. For instance, ID38 derived the
779 formula for SOP directly from its physical definition, which is specifically used for SOP calculations.
780 This direct approach is not feasible with equivalent circuit models.

781 Furthermore, while equivalent circuit models often require different estimation algorithms for
782 different state estimates, an electrochemical model enables the design of a unified battery state
783 estimation algorithm. This singular method can estimate internal battery parameters, which are then
784 used to compute various related battery states. This efficiency simplifies the overall process,
785 enhancing the functionality and accuracy of the BMS.

788 Looking forward, it is essential to conduct research on joint estimation of multiple states based on
789 electrochemical models and investigate the interaction mechanisms between different states. This will
790 fully utilize the advantages of electrochemical models.

791 792 **4.1.5 More use of internal battery information**

793 The electrochemical model provides more in-depth information about the internal workings of a
794 battery compared to the ECMs. This additional insight enables researchers to better monitor the
795 battery, optimize its performance, and ensure its safety. Not only does the electrochemical model
796 enhance the accuracy of battery state estimation, but it also plays a crucial role in safeguarding the
797 battery as well. Relying solely on the equivalent circuit model makes it challenging to meet battery
798 safety requirements, resulting in conservative designs for current BMS that do not fully exploit the
799 battery's potential.

800 As demonstrated in the results from Section 3.3, the majority of researchers still concentrate
801 primarily on battery SOC estimation. This approach fails to fully capitalize on the benefits offered by
802 electrochemical models. In ID85, the internal information calculated from the electrochemical model
803 is used to design the SOA. ID38 calculates SOP directly with internal battery information. They are
804 all good example of utilizing the battery internal information. For instance, ID46 estimates the battery
805 core temperature using an electrochemical model, effectively preventing thermal runaway. This
806 method is superior to those that rely on measuring the battery's surface temperature or calculating the
807 average temperature. Future research should explore battery safety design based on electrochemical
808 models.

809 810 **4.2 Limitations**

811 Due to the scope of this literature review being primarily focused on battery state estimation
812 based on electrochemical models, it may not comprehensively cover studies related to battery
813 modeling, model simplification, and parameter identification. Consequently, readers interested in
814 these areas may find it necessary to consult additional literature reviews or research papers within
815 those specific fields.

816 817 **5 Conclusion**

818 The widespread use of lithium-ion batteries has led to an increasing demand for enhancing battery
819 efficiency and ensuring safety. Designing battery state estimation methods based on electrochemical
820 models can optimize battery performance and safety. In this study, a systematic review of existing
821 electrochemical model-based battery state estimation research was conducted. A total of 102 research
822 articles were selected, and their research trends, electrochemical model selection, and battery state
823 estimation methods were systematically summarized. The results indicate that research in this area has
824 been growing in recent years. The results demonstrate the good results and the great potential of
825 electrochemical model-based battery state estimation achieved so far. At the same time, the gaps in
826 the current research are also raised. The majority of research has focused on individual cells, with less
827 attention paid to battery modules and packs. There is a lack of standardization in model simplification
828 methods and model parameter identification processes. Most studies have been conducted only for
829 single state estimation, with SOC being the most commonly studied, followed by SOH and SOT, and
830 very few studies have been conducted for SOP and SOB. Future research in this field should focus on
831 state estimation for modules and packs, parameter identification and model simplification for state
832 estimation, joint multi-state estimation, and utilizing more internal battery information for state
833 estimation. In conclusion, this paper demonstrates comprehensive progress in electrochemical
834 modeling-based battery state estimation. Existing problems and future research directions are also
835 evaluated and discussed, which will provide a comprehensive reference for readers interested in this
836 field.

837 838 **CRediT authorship contribution statement**

839 **Feng Guo:** Conceptualization; Data curation; Formal analysis; Investigation; Methodology;
840 Software; Validation; Visualization; Roles/Writing - original draft. **Luis Couto Mendonca:**
841 Conceptualization; Writing - review & editing. **Grietus Mulder:** Conceptualization; Writing - review

842 & editing. **Khiem Trad:** Supervision; Writing - review & editing. **Guangdi Hu:** Writing - review &
 843 editing. **Odile Capron:** Writing - review & editing. **Keivan Haghverdi:** Writing - review & editing.

844

845 **Declaration of competing interest**

846 The authors declare that they have no known competing financial interests or personal
 847 relationships that could have appeared to influence the work reported in this paper.

848

849 **Data Availability**

850 Data will be made available on request.

851

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856

857 **Appendix**

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Table A Details of the reviewed studies

ID	References	State	Model	Materials
ID1	[20]	SOC, SOH	SPM	NMC
ID2	[21]	SOH	SPM	LMO
ID3	[22]	SOC	SPM	
ID4	[23]	SOC	SPM	LFP
ID5	[24]	SOC	SPM	LCO
ID6	[25]	SOC	P2D	LMO
ID7	[26]	SOC	P2D	LFP
ID8	[27]	SOH	SPMe	LFP
ID9	[28]	SOC	SPM	LFP
ID10	[29]	SOC, SOT, SOB	SPM	
ID11	[30]	SOC, SOH, SOT	P2D	NMC
ID12	[31]	SOC, SOH	SPM	
ID13	[32]	SOC	SPMe	NMC
ID14	[33]	SOC	SPMe	LCO
ID15	[34]	SOC	SPM	LMO
ID16	[35]	SOH	P2D	
ID17	[36]	SOC	SPM	NMC
ID18	[37]	SOC, SOT	SPM	LFP
ID19	[38]	SOC, SOH	P2D	NMC
ID20	[39]	SOC, SOH	SPM	
ID21	[40]	SOC, SOH	SPMe	LCO
ID22	[41]	SOC	SPMe	LMO
ID23	[42]	SOC	P2D	NMC
ID24	[43]	SOC	P2D	
ID25	[44]	SOC	SPMe	NMC
ID26	[45]	SOC, SOH	SPMe	LMO-NMC
ID27	[46]	SOC	P2D	
ID28	[47]	SOC, SOT	SPM	LFP
ID29	[48]	SOC	SPM	LCO
ID30	[49]	SOC	SPMe	NMC
ID31	[50]	SOC	P2D	LFP
ID32	[51]	SOC	SPM	LFP

ID33	[52]	SOC	SPMe	NMC
ID34	[53]	SOC, SOH	P2D	NMC
ID35	[54]	SOC, SOH	SPM	LCO
ID36	[55]	SOC	SPM	LCO
ID37	[56]	SOC	SPM	LFP
ID38	[57]	SOP	SPM	LMO
ID39	[58]	SOH	P2D	LFP,NMC,NCA
ID40	[59]	SOC, SOH	SPM	LCO
ID41	[60]	SOC	SPM	
ID42	[61]	SOC	SPM	LFP
ID43	[62]	SOC, SOT	P2D	LCO
ID44	[63]	SOC	SPMe	LMO
ID45	[64]	SOC	P2D	
ID46	[65]	SOC, SOT	P2D	LCO
ID47	[66]	SOC, SOH	P2D	
ID48	[67]	SOB	SPM	LFP
ID49	[68]	SOC, SOT	SPM	LFP
ID50	[69]	SOC	SPM	
ID51	[70]	SOC	SPM	
ID52	[71]	SOC, SOH	SPM	LCO
ID53	[72]	SOC, SOT	P2D	LMO,NMC,NCA
ID54	[73]	SOH	SPMe	NMC
ID55	[74]	SOC	SPM	LCO
ID56	[75]	SOC	SPMe	LMO
ID57	[76]	SOH	SPM	NMC
ID58	[77]	SOC	SPMe	NMC
ID59	[78]	SOC, SOT	SPMe	LFP
ID60	[79]	SOC	P2D	LCO
ID61	[80]	SOH	P2D	LCO
ID62	[81]	SOH	SPMe	LMO-NMC
ID63	[82]	SOC	P2D	LCO
ID64	[83]	SOC	P2D	NCA
ID65	[84]	SOC	SPM	LFP
ID66	[85]	SOC	SPM	LMO
ID67	[86]	SOC	SPM	LCO
ID68	[87]	SOC	SPM	
ID69	[88]	SOC	P2D	LCO
ID70	[89]	SOC	SPM	LCO
ID71	[90]	SOC	P2D	LCO
ID72	[91]	SOC	SPM	LCO
ID73	[92]	SOC	SPM	LCO
ID74	[93]	SOC	SPMe	LFP
ID75	[94]	SOC	P2D	LFP
ID76	[95]	SOH	SPM	LMO-NMC
ID77	[96]	SOP	SPM	NMC
ID78	[97]	SOC, SOT	SPM	LCO
ID79	[98]	SOC	SPMe	LFP
ID80	[99]	SOC	SPMe	NMC
ID81	[100]	SOC	SPMe	LFP
ID82	[101]	SOH	SPM	NMC
ID83	[102]	SOC	P2D	NCA
ID84	[103]	SOC	SPMe	NMC
ID85	[104]	SOP	SPMe	NMC

ID86	[105]	SOC	P2D	LFP
ID87	[106]	SOC,SOT	SPM	LCO
ID88	[107]	SOH	SPM	LFP
ID89	[108]	SOC	P2D	LFP,NMC,NCA
ID90	[109]	SOH	P2D	
ID91	[110]	SOC	SPMe	NCA
ID92	[111]	SOC	P2D	
ID93	[112]	SOC	SPMe	NMC
ID94	[113]	SOC	SPM	LCO
ID95	[114]	SOC	SPM	
ID96	[115]	SOH	SPM	LFP
ID97	[116]	SOP	P2D	NMC
ID98	[117]	SOC	SPM	LCO
ID99	[118]	SOC	P2D	NMC
ID100	[119]	SOC	SPM	NMC
ID101	[120]	SOH	SPMe	LFP
ID102	[121]	SOC	SPMe	LCO

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